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A New Approach to (+)-Brefeldin A via a Nickel-catalyzed Coupling Reaction of Cyclopentenyl Acetate and Lithium 2-Furylborate

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Abstract: Coupling reaction of 2 with 2-furylborate 14 in the presence of the nickel catalyst afforded 1,4-isomer 5 regio- and stereoselectively and the product 5 was applied to the synthesis of Bartlett's brefeldin intermediate 11. Copyright © 1996 Elsevier Science Ltd

Transition metal-catalyzed coupling reactions of cyclopentadiene monoepoxide (1) or its synthetic equivalent such as 2 with "hard" organometallic reagents, especially those possessing metal-sp²-carbon bond is an attractive processes for the synthesis of biologically important cyclopentanoids such as hitachimycin, hybridalactone, brefeldins, multifiden etc. (eq. 1). Among the organometallic/metal-catalyst systems published so far for coupling with acyclic allylic derivatives, 1 organotin/Pd is the only one which has been applied to cyclopentenyl derivatives. 2 Although perfect stereocontrol is obtained in the reaction of 1 with all three organotins examined, the observed low regioselectivity suggests difficulty of regiocontrolled installation of "hard" species onto the 5-membered ring system.3

O'...OAc
$$\frac{R-m}{Ni(0)}$$
 $\left[\bigcirc \bigcap_{H} \bigcap_{Ni-H} H \right]$ 1,4-isomer and/or 1,2-isomer 1,2-isomer

Recently we have reported regio- and stereoselective coupling reactions of secondary allylic carbonates and lithium organoborates in the presence of nickel catalysts (eq. 2).⁴ We expected that due to the small radius of nickel, the nickel atom on intermediate 3 is closely bound to the π -allyl moiety and, consequently, that transmetallation and subsequent reductive elimination processes would become more susceptible to a small steric hindrance and/or an electronic effect, thus producing the 1,4-isomer regioselectively. We selected brefeldin A (4)⁵ as the target molecule and examined such an effect in the asymmetric synthesis of the new intermediate 5 where the furyl group is chosen for the C_1 - C_4 (brefeldin A numbering) moiety of 4. Since furans are stable

$$R^{1} \longrightarrow R^{2} \qquad \frac{R^{3} - B(OMe)_{3}Li}{NiCl_{2}Ln} \qquad R^{1} \longrightarrow R^{2}$$

$$R^{1} = Ph, COOEt \qquad R^{3} = Ph, 2 \cdot Furyl.$$
Alkenyl

under basic, weakly acidic, and mild oxidative conditions, a number of reactions are operative for installation of the C_{10} - C_{16} side chain and functional group transformations. Herein we report the regio- and stereoselective formation of optically active 5 and its conversion to Bartlett's intermediate 11^{5c} (Scheme 1).

Although epoxide 1^6 and acetate 2^7 are readily available as optically active compounds, preliminary experiments were carried out with racemic substrates $(rac-1, rac-2)^8$ and 3 equiv of lithium 2-furylborate 14 (Fu = 2-furyl) in the presence of 5-10 mol% of the nickel catalyst (NiCl₂(dppf) or NiCl₂(PPh₃)₂) under the conditions reported (THF, 50-60 °C, 4-12 h). ^{4a} When the epoxide rac-1 was submitted to the reaction with NiCl₂(PPh₃)₂, all four possible isomers rac-5, $15-17^9$, 10 were produced in a ratio of 42:33:23:2 (determined by 300 MHz 1 H NMR) and in low yield (<30%) (eq. 3). No other conditions (NiCl₂(dppf), room temp., or Et₂O) improved the selectivity and yield. On the other hand, the acetate rac-2 did afford the trans isomers, rac-5 and -15, in a ratio of 88:12 and in 78% yield with NiCl₂(PPh₃)₂ (eq. 4). A similar ratio (83:17) was observed with NiCl₂(dppf) (55% combined yield). In the 1 H NMR spectra of the crude products, no cis isomer(s) were detected. It should be noted that this is the first example to realize practically high levels of regions election in the coupling of "hard" organometallics and π -allyl metals derived from cyclopentenyl compounds. 11 The coupling reaction was carried out again with optically active 2^{7a} ([α]²⁸ $_D$ = -66 (c 0.82, CHCl₃); for 2 of >96% ee, 7a [α]²⁰ $_D$ = -68 (CHCl₃)) and borate 14 to afford 5 ([α]²⁹ $_D$ = +190 (c 0.46, CHCl₃)) and 15 ([α]²⁹ $_D$ = -148 (c 0.15, CHCl₃)), which were separated easily by silica gel column chromatography.

$$rac-1 \xrightarrow{14} \begin{array}{c} \text{HO...} \\ \hline 14 \\ \text{Ni cat.} \\ \hline \\ rac-2 \\ \hline \\ \text{NiCl}_2(\text{PPh}_3)_2 \\ \hline \\ \text{NiCl}_2(\text{dppf}) \end{array} \begin{array}{c} \text{Fu} \\ \hline \\ \\ \text{Fu} \\ \hline \\ \\ \text{Fu} \\ \hline \\ \text{Fu} \\ \\ \text{Fu}$$

The optically active side chain 12 ($[\alpha]^{20}_D = +14$ (c 0.99, Et₂O); the calculated value for the pure enantiomer, 5d [α] $^{20}_D = +12.6$ (c 1.04, Et₂O)) was prepared from (R)-(-)-epichlorohydrin (18) (98.8% ee) by the route shown below. Hydrozirconation of acetylene 20 was conveniently carried out by the method of Lipshutz. 12

Scheme 1. (a) PCC, CH₂Cl₂; (b) 13 (12, n-BuLi then LiCu(CN)Th), -78 °C, THF; (c) LiB(s-Bu)₃H, THF, -78 °C; (d) HCOOH, DEAD, PPh₃ then NaOH; (e) MOMCl, i-Pr₂NEt; (f) Bu₄NF, THF; (g) NBS, C₅H₅N, acetone/H₂O, -20 °C, 30 min then rt., 4 h; (h) NaClO₂, resorcinol, i-BuOH, buffer (pH 3.6).

Oxidation of 5 with PCC proceeded without affecting the furan ring^{13a} to afford cyclopentenone 6 ($[\alpha]^{30}_D = +280 \ (c\ 0.08, \text{CHCl}_3)$) in 71% yield (Scheme 1). 1,4-Addition of the higher order cuprate 13 (1.5 equiv) derived from 12 [(1) *n*-BuLi, -70 °C, 1 h; (2) LiCu(CN)(2-Th), ¹⁴ -70 °C] provided cyclopentanone 7 quantitatively. Reduction of 7 using LiB(s-Bu)₃H followed by Mitsunobu inversion furnished 8 and the C₇ epimer in a ratio of 3:1. Protection (MOMCl, *i*-Pr₂NEt) followed by desilylation (Bu₄NF) gave alcohol 9 in 83% yield.

Conversion of the furan ring of 9 into the 4-oxo-2-butenoic acid moiety was then explored. Although attempts to apply the literature protocols (PCC/CH₂Cl₂, ^{13a,b} Br₂/aq. acetone ^{13c} or aq. MeCN^{13d}) were all in vain in our hand, NBS in aq. acetone was found to be quite effective, furnishing the aldehyde 10 in 71% yield. Further oxidation of 10 with NaClO₂ in the presence of resorcinol ¹⁶ provided the acid 11 in 78% yield, whose ¹H NMR spectrum was in good agreement with that reported by Bartlett. ^{5c}

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